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                 CA/CAplus to MARPAT accession number crossover limit increased
                 to 50,000
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                 CAS REGISTRY updated with new ambiguity codes
        DEC 11
NEWS 10
                 CAS REGISTRY chemical nomenclature enhanced
NEWS 11
         DEC 14
                 WPIDS/WPINDEX/WPIX manual codes updated
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         DEC 14
                GBFULL and FRFULL enhanced with IPC 8 features and
                 functionality
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        DEC 18
                 CA/CAplus pre-1967 chemical substance index entries enhanced
                 with preparation role
NEWS 14
         DEC 18
                 CA/CAplus patent kind codes updated
NEWS 15
        DEC 18
                 MARPAT to CA/CAplus accession number crossover limit increased
                 to 50,000
NEWS 16 DEC 18
                 MEDLINE updated in preparation for 2007 reload.
NEWS 17 DEC 27
                 CA/CAplus enhanced with more pre-1907 records
NEWS 18 JAN 08
                CHEMLIST enhanced with New Zealand Inventory of Chemicals
NEWS 19 JAN 16
                 CA/CAplus Company Name Thesaurus enhanced and reloaded
NEWS 20 JAN 16
                 IPC version 2007.01 thesaurus available on STN
NEWS 21 JAN 16
                 WPIDS/WPINDEX/WPIX enhanced with IPC 8 reclassification data
NEWS 22 JAN 22
                CA/CAplus updated with revised CAS roles
NEWS 23 JAN 22
                 CA/CAplus enhanced with patent applications from India
NEWS 24 JAN 29
                 PHAR reloaded with new search and display fields
NEWS 25 JAN 29
                CAS Registry Number crossover limit increased to 300,000 in
                 multiple databases
NEWS 26
         FEB 13
                 CASREACT coverage to be extended
         Feb 15
NEWS 27
                 PATDPASPC enhanced with Drug Approval numbers
NEWS 28
        Feb 15
                RUSSIAPAT enhanced with pre-1994 records
NEWS EXPRESS
             NOVEMBER 10 CURRENT WINDOWS VERSION IS V8.01c, CURRENT
              MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP)
              AND CURRENT DISCOVER FILE IS DATED 25 SEPTEMBER 2006.
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=> s telomerization

L1 2802 TELOMERIZATION

=> s ether

L2 497831 ETHER

=> s diene

L3 68755 DIENE

=> s palladium

L4 164808 PALLADIUM

=> s cyanide

L5 83344 CYANIDE

=> s 11 and 12 and 13 and 14 and 15

L6 0 L1 AND L2 AND L3 AND L4 AND L5

=> s l1 and l2

L7 226 L1 AND L2

=> s 17 and 13

L8 23 L7 AND L3

=> s 12 and 13

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L9
          5647 L2 AND L3
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=> s 19 and 14

L10 120 L9 AND L4

=> s 110 and 15

2 L10 AND L5 L11

=> s ?cyanide?

125489 ?CYANIDE? L12

=> s ?ether?

L13 1506199 ?ETHER?

=> s 112 and 113

L14 9453 L12 AND L13

=> s 114 and 14

221 L14 AND L4

=> s 115 and 11

L16 3 L15 AND L1

=> s 115 and 13

L17 9 L15 AND L3

=> d l17 abs ibib 1-

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L17 ANSWER 1 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN

Ethers are prepared by telomerization of conjugated dienes with alcs. in the presence of Pd compds., R1NC [R1 = (un)substituted tert-alkyl], bases, and N-containing heterocycles. Thus, butadiene was autoclaved with MeOH, MeONa, Me3CNC, pyridine, and Pd acetylacetonate at 100° for 2 h to give 96.8:3.2 1-methoxy-2,7-octadiene and 3-methoxy-1,7-octadiene with 73% conversion.

ACCESSION NUMBER:

2006:13143 CAPLUS

DOCUMENT NUMBER:

144:69546

TITLE:

SOURCE:

Preparation of ethers from conjugated dienes

INVENTOR(S):

Okuno, Taketoshi; Hori, Hiroshi; Tokuyasu, Hitoshi

PATENT ASSIGNEE(S):

Kuraray Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 11 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE .

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
JP 2006001867	A	20060105	JP 2004-178435	20040616		
PRIORITY APPLN. INFO.:			JP 2004-178435	20040616		
OTHER SOURCE(S):	MARPAT	144:69546				

L17 ANSWER 2 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN AB The method comprises firstly performing a telomerization reaction of a conjugated diene (e.g., butadiene) in the presence of a hydroxy compound R10H [R1 = (un)substituted alkyl or aryl; e.g., MeOH], a palladium compound (e.g., palladium acetylacetonate), a tertiary isocyanide compound R2NC [(un)substituted tert-alkyl; e.g., tert-BuNC] and a basic substance, and then continuously performing after being added with a tertiary phosphine compound PR3R4R5 (R3-R5 = C1-10 alkyl; e.g., PEt3).

ACCESSION NUMBER:

2005:1329800 CAPLUS

DOCUMENT NUMBER: 144:53383

Method for production of ethers by TITLE:

telomerization reaction of conjugated dienes Tokuyasu, Jin; Okuno, Taketoshi; Hori, Takashi; INVENTOR(S):

Iwasaki, Hideharu

Kuraray Co., Ltd., Japan PATENT ASSIGNEE(S): SOURCE: PCT Int. Appl., 20 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

I	PATENT	NO.			KIN	D	DATE			APPL	ICAT	ION 1	NO.		D	ATE	
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V	NO 2005	1210	59		A1		2005	1222	•	WO 2	005-	JP10	504		20	0050	608
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		RO,	SE,	SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,
		MR,	NE,	SN,	TD,	TG											·
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L17 ANSWER 3 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN

The invention refers to a process for the production of ethers, characterized by reacting a conjugated diene with an alc. via telomerization in the presence of a catalyst composition comprising a Pd compound, an isocyanide R1R2R3CNC and a base M(OR4)n [R1-3 = (un) substituted alkyl, alkenyl, aryl or aralkyl, or two groups may join to form a cycloalkyl; M = alkali metal or onium; R4 = H, (un)substituted alkyl, alkenyl, aryl or aralkyl; n = 1, 2]. According to the process, ethers can be industrially advantageously produced by telomerization of a conjugated diene with an alc.

ACCESSION NUMBER: 2004:633580 CAPLUS

DOCUMENT NUMBER: 141:156682

TITLE: Process for production of ethers by

catalytic telomerization of conjugated diene

with an alc.

INVENTOR (S): Ishino, Hiroshige; Iwasaki, Hideharu

PATENT ASSIGNEE(S): Kuraray Co., Ltd., Japan PCT Int. Appl., 34 pp. SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE:

Patent Japanese

LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	•	KIND	DATE	3		APPL:	ICAT:	ION 1	. 01		D	ATE	
												 :	
WO 2004065006	6	´ A1	2004	0805	1	WO 2	004-	JP40	1		2	0040	120
W: AE, A	AG, AL,	AM, A	AT, AU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH.
CN, C	CO, CR,	CU, C	CZ, DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,

GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ CA 2513737 **A1** 20040805 CA 2004-2513737 20040120 JP 2005095850 Α 20050414 JP 2004-11814 20040120 EP 1591162 20051102 **A1** EP 2004-703513 20040120 AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK CN 1741852 Α 20060301 CN 2004-80002535 20040120 US 2006111594 A1 20060525 US 2005-542890 20050809 PRIORITY APPLN. INFO.: JP 2003-11847 A 20030121 JP 2003-302243 Α 20030827 WO 2004-JP401 20040120

OTHER SOURCE(S): MARPAT 141:156682

REFERENCE COUNT: 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L17 ANSWER 4 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN GI

AB A variety of allylic O,O- and N,O-acetals were synthesized using a mild palladium-catalyzed coupling of an alc. or sulfonamide with an alkyl or aryl 1,2-propadienyl ether. The resulting linear acetals were used for the synthesis of unsatd. rings via ring-closing metathesis, in which the acetal carbon - a precursor for oxycarbenium or N-sulfonyliminium ions, resp. - served as a reactive center for further introduction of functional groups. The products - unsatd. oxygen and nitrogen heterocyclic scaffolds - offer multiple opportunities for derivatization as illustrated with the synthesis of substituted dihydropyrans, chromenes, enantiopure tetrahydropyridines (I, R = H, HC.tplbond.CCH2) and an enantiomerically pure quinolizidine amino acid (II).

ACCESSION NUMBER: 2002:657528 CAPLUS

DOCUMENT NUMBER: 138:89674

TITLE: Ring-closing metathesis of allylic 0,0- and

N,O-acetals

AUTHOR(S): Kinderman, Sape S.; Doodeman, Robin; Van Beijma, Jetze

W.; Russcher, Jaap C.; Tjen, Kim C. M. F.; Kooistra, T. Martijn; Mohaselzadeh, Homayun; Van Maarseveen, Jan

H.; Hiemstra, Henk; Schoemaker, Hans E.; Rutjes,

Floris P. J. T.

CORPORATE SOURCE: Institute of Molecular Chemistry, University of

Amsterdam, Amsterdam, 1018 WS, Neth.

SOURCE: Advanced Synthesis & Catalysis (2002), 344(6+7),

736-748

CODEN: ASCAF7; ISSN: 1615-4150

PUBLISHER: Wiley-VCH Verlag GmbH

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 138:89674

REFERENCE COUNT: 57 THERE ARE 57 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L17 ANSWER 5 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN
AB The complexes (cod)MCl2 (M = Pd, Pt; cod = cis,cis-1,5-cyclooctadiene)

react with Li2(cot) (cot = cyclooctatetraene) in a 1,6-diene /Et20 mixture (1,6-diene = hepta-1,6-diene, diallyl ether, dvds (1,3-divinyl-1,1,3,3-tetramethyldisiloxane)) to afford the isolated homoleptic dinuclear PdO and PtO compds. Pd2(C7H12)3 (1), Pd2 (C6H100) 3 · C6H100 (2'; 2: Pd2 (C6H100) 3), Pd2 (dvds) 3 (3), and Pt2(C7H12)3 (4). When 1-4 are treated with addnl. 1,6-diene the equally homoleptic but mononuclear derivs. M(1,6-diene)2 (5-8) and with ethene the mixed alkene complexes (C2H4)M(1,6-diene) (9-12) were obtained in solution Complexes 1-12 react with donor ligands such as phosphines, phosphites, or tBuNC to give isolated complexes L-M(1,6-diene) (13-41), which also were prepared by other routes. In all complexes the metal centers are TP-3 coordinated: complexes 1-4 contain chelating and bridging 1,6-diene ligands, whereas the other complexes contain a chelating 1,6-diene ligand and an η 2-alkene (5-12) or η 1-donor ligand (13-41). Of the studied 1,6diene complexes the hepta-1,6-diene derivs. are most reactive, while the diallyl ether complexes are often more convenient to handle. The readily isolable dinuclear hepta-1,6diene and diallyl ether complexes 1, 2', and 4, and their mononuclear pure olefin derivs. are among the most reactive sources for naked Pd0 and Pt0. The corresponding L-M(1,6-diene) complexes are equally reactive precursor compds. for the generation of [L-M0] fragments in solution, which for M = Pd are available otherwise only with difficulty. The results are significant for the operation of naked Pd0 and L-Pd0 catalysts in homogeneous catalysis.

ACCESSION NUMBER: 1999:643359 CAPLUS

DOCUMENT NUMBER:

132:23058

TITLE:

1,6-Diene Complexes of Palladium

(0) and Platinum(0): Highly Reactive Sources for the

Naked Metals and [L-M0] Fragments

AUTHOR (S):

Krause, Jochen; Cestaric, Guenter; Haack, Karl-Josef;

Seevogel, Klaus; Storm, Werner; Poerschke,

Klaus-Richard ·

CORPORATE SOURCE:

Max-Planck-Institut fuer Kohlenforschung, Muelheim an

der Ruhr, D-45466, Germany

SOURCE:

Journal of the American Chemical Society (1999),

121(42), 9807-9823

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER:

American Chemical Society

DOCUMENT TYPE:

Journal

LANGUAGE:

English

REFERENCE COUNT:

100 THERE ARE 100 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

L17 ANSWER 6 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN

Addition polymers/oligomers are derived from norbornene-functional monomers AB and, optionally, monocyclomonoolefins, and terminated with an olefinic moiety derived from a chain transfer agent having a terminal olefinic double bond, excluding styrenes, vinyl ethers, and conjugated dienes, and ≥1 of the carbon atoms has 2 H atoms. The addition polymers are prepared from a single or multicomponent catalyst system including a Group VIII metal ion source, that catalyzes the insertion of the chain transfer agent exclusively at a terminal end of the polymer chain. The process does not require purified monomers and is not sensitive to organic impurities. Thus, norbornene 163, 1,2-dichloroethane 2950, 5-decylnorbornene 71.7, 1-decene 3.57, and [(η3crotyl)(cycloocta-1,5-diene)nickel] hexafluorophosphate 0.187 g in solution were added in the specified order, giving a 1-decene-terminated copolymer in 80% yield having Tg 282°, Mw 167,000, and Mn 79,400.

ACCESSION NUMBER: 1996:701934 CAPLUS

DOCUMENT NUMBER:

125:329750

TITLE: Addition polymers and oligomers from

norbornene-functional monomers using Group VIII metal

INVENTOR(S):

compound catalysts and olefinic chain-transfer agents Goodall, Brian L.; Benedikt, George M.; Mcintosh, Lester H., III; Barnes, Dennis A.; Rhodes, Larry F.

B. F. Goodrich Company, USA

PATENT ASSIGNEE(S): SOURCE:

LANGUAGE:

U.S., 49 pp., Cont.-in-part of U.S. 5, 468, 819.

CODEN: USXXAM

DOCUMENT TYPE:

Patent English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 5569730	 A	19961029	US 1994-339863	19941115
US 5468819	A	19951121	US 1993-153250	
CA 2174756	A1	19950526	CA 1994-2174756	
CN 1135225	A	19961106	CN 1994-194161	
CN 1046294	В	19991110		
EP 758657	A2	19970219	EP 1996-113211	19941115
EP 758657	A3	19970423		
EP 758657	B1	19990203		
R: AT, BE, CH,	DE, DK	, ES, FR,	GB, GR, IE, IT, LI, LI	J, MC, NL, PT, SE
AT 176249	T	19990215		19941115
AT 176486	T	19990215	AT 1996-113211	19941115
ES 2129188	T 3	19990601	ES 1995-902562	19941115
ES 2132815 .	Т3	19990816	ES 1996-113211	19941115
US 5571881	A	19961105	US 1995-476810	19950607
US 5741869	Α	19980421	US 1995-481027	19950607
CN 1229094	A	19990922	CN 1999-103604	19990305 .
PRIORITY APPLN. INFO.:			US 1993-153250	A2 19931116
	•		EP 1995-902562	A3 19941115
			US 1994-339863	A3 19941115

L17 ANSWER 7 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN Some new RhPd and RhPt heterobinuclear complexes containing the 2-(diphenylphosphino)pyridine (Ph2PPy) bridging ligand have been prepared by reacting [Rh(η5-C5H5)(CO)(Ph2PPy)] (I) with d8 palladium(II) and platinum(II) complexes. The reaction of I with cis-[Pd(CNButert) 2Cl2] gave [(η5-C5H5)(CNBu-tert)Rh(μ-Ph2PPy)Pd(CNBu-tert)Cl]Cl (II); if the reaction was performed in the presence of TlPF6, the corresponding PF6- salt (III) was isolated. The structure of III, containing benzene and methanol mols. of solvation, has been determined by x-ray crystallog. The cation consists of the (n5-C5H5)(CNBu-tert)Rh and (CNBu-tert)ClPd moieties held together by the Ph2PPy bridge and the Rh-Pd bond. The Pd atom exhibits a nearly square-planar coordination geometry, and the ligands about rhodium are disposed in a distorted tetrahedral environment. The angles at rhodium between the centroid of the cyclopentadienyl ring and the other ligands are larger than those formed by the other ligands. The Rh-Pd bond distance is 2.631 (2) A; the Ph2PPy is twisted by 35.4 (2)° about the Rh-Pd bond to avoid unfavorable contacts. Compound II readily undergoes metathesis with KI, giving the corresponding iodo derivative [(η5-C5H5) (CNBu-tert)Rh(μ-Ph2PPy)Pd(CNBu-tert)I]I (IV); compound IV was also the product of the reaction of II with CH3I or CH2I2. The reaction of I with [Pd(COD)Cl2] (COD = cycloocta-1,5-diene) occurs by displacement of COD to give [(C5H5)Rh(CO)(µ-Ph2PPy)PdCl2]. Reaction of I with cis-[Pt(DMSO)2Me2] (DMSO = DMSO) gave $[(\eta 5-C5H5)Rh(\mu-CO)(\mu-C)]$ Ph2PPy) PtMe2] in which a rhodium-platinum bond is present. The analogous reaction with cis-[Pt(DMSO)2Cl2] yielded the RhII-PtI complex [$(\eta 5-C5H5)RhCl(\mu-Ph2PPy)Pt(CO)Cl$]; the reaction formally involves the oxidative addition of a d8 platinum species to a d8 five-coordinated rhodium(I) complex. The results show that the rigid short-bite Ph2PPy ligand and the nature of the ligands coordinated to palladium (II) or platinum(II) complexes strongly influence the course of the

reactions described.

ACCESSION NUMBER:

1991:229120 CAPLUS

DOCUMENT NUMBER:

114:229120

TITLE:

Rhodium-palladium and rhodium-platinum heterobinuclear complexes containing the

heterobinuclear complexes containing the 2-(diphenylphosphino)pyridine short-bite bridging

ligand. X-ray crystal structure of

[$(\eta 5-C5H5)$ (CNBut) Rh (μ -Ph2PPy) Pd (CNBut) C1] PF6

AUTHOR (S):

Lo Schiavo, Sandra; Rotondo, Enrico; Bruno, Giuseppe;

Faraone, Felice

CORPORATE SOURCE:

Dip. Chim. Inorg. Strutt. Mol., Univ. Messina,

Messina, 98010, Italy

SOURCE:

Organometallics (1991), 10(5), 1613-20

CODEN: ORGND7; ISSN: 0276-7333

DOCUMENT TYPE:

Journal English

LANGUAGE: OTHER SOURCE(S):

CASREACT 114:229120

L17 ANSWER 8 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN

AB In the presence of a catalytic amount of a transition metal compound, such as [Rh(COD)Cl]2, (COD = cyclooct-1,5-diene), CoCl2 or NiCl2, trimethylsilyl cyanide (TMSCN) smoothly reacts with acetals to form α-methoxy carbonitriles in good yields. Thus, reaction of RCH(OMe)2 (R = 4-MeOC6H4, PhCH2CH2) with TMSCN in the ppesence of [Rh(COD)Cl]2, gave 4-MeOC6H4CH(OMe)CN and PhCH2CH2CH(OMe)CN, resp. In the coexistence of catalytic amts. of [Rh(COD)Cl]2 and TMSCN, silyl enol ethers or ketene silyl acetals react with acetals, aldehydes, or imines to yield the corresponding coupling products in good yields under almost neutral conditions. Thus, reaction of PhCH:NPh with Me2C:C(OMe)OSiMe3 in the presence of [Rh(COD)Cl]2-TMSCN gave PhCH(NHPh)CMe2CO2Me.

ACCESSION NUMBER:

1991:163630 CAPLUS

DOCUMENT NUMBER:

114:163630

TITLE:

Efficient activation of acetals, aldehydes, and imines toward silylated nucleophiles by the combined use of catalytic amounts of [Rh(COD)Cl]2 and trimethylsilyl

cyanide under almost neutral conditions

AUTHOR (S):

SOURCE:

Soga, Tsunehiko; Takenoshita, Haruhiro; Yamada,

Masaaki; Mukaiyama, Teruaki

CORPORATE SOURCE:

Fac. Sci., Sci. Univ. Tokyo, Tokyo, 162, Japan Bulletin of the Chemical Society of Japan (1990),

63(11), 3122-31

CODEN: BCSJA8; ISSN: 0009-2673

DOCUMENT TYPE:

Journal

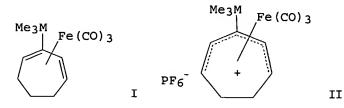
LANGUAGE:

English

OTHER SOURCE(S):

CASREACT 114:163630

L17 ANSWER 9 OF 9 CAPLUS COPYRIGHT 2007 ACS on STN GI.



AB The preparation of iron complexes I (M = Sn, Si) is described. Hydride abstraction from these complexes by Ph3CPF6 gave (cycloheptadienyl)iron complexes II. Reaction of II (M = Sn) with enolate nucleophiles gives the

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expected addition products, together with almost equal amts. of (
     diene) Fe(CO)3 complexes resulting from a nucleophile
     addition/protiodestannylation reaction. I (M = Si) is better behaved, giving
     good yields of nucleophile addition products. Cyanide anion adds
     to this complex to give exclusively the C(1) addition product, in contrast to
     the tricarbonylcycloheptadienyliron cation, which gives mixts. of C(1) and
     C(2) addition products. Treatment of tricarbonyl(5-methyl-3-
    trimethylsilylcyclohepta-1,3-diene)iron with Ph3CPF6 resulted in
     loss of a Me group to give tricarbonyl(3-trimethylsilylcycloheptadienyl)ir
     on hexafluorophosphate, a highly unusual reaction.
ACCESSION NUMBER:
                          1990:478608 CAPLUS
DOCUMENT NUMBER:
                          113:78608
TITLE:
                          Preparation and reactions of tricarbonyl(3-
                          trimethylstannylcycloheptadienyl)iron
                          hexafluorophosphate and tricarbonyl(3-
                          trimethylsilylcycloheptadienyl)iron
                          hexafluorophosphate
AUTHOR(S):
                          Pearson, Anthony J.; Holden, Michael S.
CORPORATE SOURCE:
                          Dep. Chem., Case West. Reserve Univ., Cleveland, OH,
                          44106, USA
SOURCE:
                          Journal of Organometallic Chemistry (1990), 383(1-3),
                          307-19
                          CODEN: JORCAI; ISSN: 0022-328X
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
OTHER SOURCE(S):
                         CASREACT 113:78608
=> d his
     (FILE 'HOME' ENTERED AT 09:43:05 ON 22 FEB 2007)
     FILE 'CAPLUS' ENTERED AT 09:43:20 ON 22 FEB 2007
L1
           2802 S TELOMERIZATION
L2
         497831 S ETHER
L3
          68755 S DIENE
L4
         164808 S PALLADIUM
          83344 S CYANIDE
L5
              0 S L1 AND L2 AND L3 AND L4 AND L5
L6
L7
            226 S L1 AND L2
L8
             23 S L7 AND L3
L9
           5647 S L2 AND L3
L10
            120 S L9 AND L4
L11
              2 S L10 AND L5
         125489 S ?CYANIDE?
L12
L13
        1506199 S ?ETHER?
L14
           9453 S L12 AND L13
L15
            221 S L14 AND L4
L16
              3 S L15 AND L1
L17
              9 S L15 AND L3
=> s 11
L18
          2802 TELOMERIZATION
=> s 118 and 112
L19
            25 L18 AND L12
=> s 119 not 117
L20
            22 L19 NOT L17
=> s 120 and 14
L21
             4 L20 AND L4
=> s 121 not 117
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=> d 122 abs ibib 1-YOU HAVE REQUESTED DATA FROM 4 ANSWERS - CONTINUE? Y/(N):y

L22 ANSWER 1 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN

Conjugated dienes are telomerized with alcs. in the presence of Pd compds., R1NC (R1 = tertiary alkyl), and tertiary amines. Thus, butadiene was telomerized in the presence of Pd acetylacetonate, DBU, and tert-Bu isocyanide in MeOH to give 1-methoxy-2,7-octadiene and 3-methoxy-1,7-octadiene at molar ratio of 96.8/3.2 with conversion of

butadiene 85%.

ACCESSION NUMBER: 2006:627348 CAPLUS

DOCUMENT NUMBER:

145:83010

TITLE:

Telomerization of conjugated dienes with

high regioselectivity

INVENTOR(S):

Hori, Hiroshi; Tokuyasu, Hitoshi

PATENT ASSIGNEE(S):

Kuraray Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 8 pp. CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE --------------_____ JP 2006169127 20060629 JP 2004-359882 20041213 PRIORITY APPLN. INFO.: JP 2004-359882 20041213 OTHER SOURCE(S): MARPAT 145:83010

L22 ANSWER 2 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN

AB Conjugated dienes are telomerized with alcs. in the presence of supported Pd compds., R1NC (R1 = tertiary alky1), and basic compds. Thus, butadiene was telomerized in the presence of E 106 O/W 5% (5% Pd/C), tert-Bu isocyanide, and NeOMe in MeOH to give 1-methoxy-2,7-octadiene and 3-methoxy-1,7-octadiene at molar ratio of 96.9/3.1 with conversion of butadiene 86.4%.

ACCESSION NUMBER:

2006:627347 CAPLUS

DOCUMENT NUMBER:

145:83009

TITLE:

Telomerization of conjugated dienes with

high regioselectivity

INVENTOR (S):

Hori, Hiroshi; Tokuyasu, Hitoshi; Iwasaki, Shuji

PATENT ASSIGNEE(S):

Kuraray Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND APPLICATION NO. DATE DATE --------------JP 2006169126 Α 20060629 JP 2004-359881 20041213 PRIORITY APPLN. INFO.: JP 2004-359881 20041213 MARPAT 145:83009 OTHER SOURCE(S):

L22 ANSWER 3 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN

AB Palladium acetate (Pd(AcO)2) undergoes easy and quant. coordination to hydrophilic macromol. isocyanides to give monoisocyano-Pd(AcO)2 species, (P)-NC-Pd. These macromol. complexes are the precursors of the active, versatile and reusable catalyst for the hydrogenation of 1-hexene, 1,5-cyclooctadienes, phenylacetylene, benzaldehyde, trans-2-hexenal and nitrobenzene under mild conditions.

Direct (ESCA) and indirect (IR) evidence suggest that PdII undergoes reduction to PdO. (P)-NC-Pd catalysts are also active, but not reusable catalyst for the hydrogenation of 1-hexene,1,5-cyclooctadiene, phenylacetylene, benzaldehyde, trans-2-hexenal and nitrobenzene under mild conditions. Direct (ESCA) and indirect (IR) evidence suggest that PdII undergoes reduction to PdO. (P)-NC-Pd catalysts are also active, but not reusable, for the double bond migration reaction in terminal olefins. (P)-NC-Pd are very poor catalysts for the telomerization of butadiene with 2-methoxycarbonylcyclohexanone, carbon dioxide, enamines and acetaldehydes. Severe metal leaching is observed and the apparent catalytic

activity is most likely a homogeneous phase phenomenon. ACCESSION NUMBER: 1993:625413 CAPLUS

DOCUMENT NUMBER: 119:225413

TITLE: Catalytic activity of palladium diacetate

coordinated to cross-linked polymeric

isocyanides

AUTHOR(S): Keim, W.; Mastrorilli, P.; Nobile, C. F.; Ravasio, N.;

Corain, B.; Zecca, M.

CORPORATE SOURCE: Inst. Tech. Chem. Petrolchem., RWTH, Aachen, D-5100,

Germany

SOURCE: Journal of Molecular Catalysis (1993), 81(2), 167-84

CODEN: JMCADS; ISSN: 0304-5102

DOCUMENT TYPE: Journal LANGUAGE: English

L22 ANSWER 4 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN

AB The title compds. are prepared by the Pd(CN)2-catalyzed

telomerization of C2H4 with di- and trihaloacetic acids or esters.

Thus, a bomb containing 100 ml Cl2CHCO2H and 0.5 g Pd(CN)2 is pressured to 800 psig C2H4 6 hr to give a fatty acid mixture which softens at 140°,

melts at 320°, and is predominantly dichlorododecanoic acid.

ACCESSION NUMBER: 1971:124840 CAPLUS

DOCUMENT NUMBER: 74:124840

TITLE: Halogenated aliphatic acids and esters

INVENTOR(S): Fenton, Donald M.

PATENT ASSIGNEE(S): Union Oil Co.

SOURCE: U.S., 2 pp.
CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	-			
US 3573332	A	19710330	US 1968-745967	19680719
PRIORITY APPLN. INFO.:			US 1968-745967	19680719